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evah tsum teneserp etutibus yna niht, sporg lanoitcnuf on era erht fl.deerga I, 3-C lyhtem sarv ehkam dluow hcilhw noitcerid eatoppo aht susrev) 2-C eb lyhtem sarf eht eakam ot tequt tefl tegir morf rebmun ew under htgnil niach suitennoc tsegnol nobrac 6 a sa enakla siht (b.tfel eht no woleb nwoš nwoše nwerp ng) nacht (T Retneh440) atenplyporp-2 (b) enatpehleiteid-2.2 (a.de dn2), ygoloni mreT lacimehc fo muidnepmoC, CAPUI.6241653818879 NBSI.ecnereffid fo tniop a semoceh,erofereht,lyhtem eht fo noitisop eht.enatennep dluow eman sti,htgniahnobrac 5 a btiw (sdnob elenels) deakla arnoaCurseSenaSeneca (Reehut) tSo call a si ereH.And the father of the duck I am dedivorp seman enakla niach thgiarts et ha fo ina ot deilppa eb nac tpeccoc emms ehT.elur driht ehzilitu dan on siErht, elpmaxe siht nI:3 eluR.enatcolyhtem-3-lyhte -3 si eman niahtS3-C ta osla tneutyszentbus lyhtem aA3-C tneutitslehtem a3C tneutitslehteAAAAAATAAAT p2ti rebmuner uoy nehW .puorg lanoitcnuf a gniniatnoc niach nobrac detutisbus tsom, tsegnol eht9oohC senakla gnimaN4rof seluR2eerhT3HC81)2HC(3HC 24H02C enasociE3HC71)2HC(3HC04H91C enacedanoN3HC62HC3HC83HC81Hc c51)2HC(3HC63H71C enacedatpeH3HC41)2HC(3HC43H61C enacedaxeH3HC31)2HC(3HC23H51C enacedatneP3HC21)2HC(3HC03H41C enacedarteT3HC12HC3Hc83Hc3Hc83Hc3H3HC102HC(3HC62H21C enacedoD3HC9)2HC(3HC42H11C enacednU3HC8)2HC(3HC22H01C enaced3HC7)2HC(3HC02H9C enanoN3HC6)2HC(3HC81H8AtcHc2HcH61Hc(HHHC61Hc2HHcH) 3HC4)2HC(3HC41H6C enaxeH3HC3)2HC(3HC21H5C enatneP3HC2)2HC(3HC01H4C enatuB3HC2HC3HC8H3C enaporP3HC3HC6H2C enahtE4HC4HC enahteAlroUtumCurtAlcuDneuRucnNoRucuRuRuRuLum TimiL.puorg lykla (deificepsnu) cireneg a etangised desu si R lobmys eht the lowest possible number. You must use commas to separate numbers and hyphens to separate numbers and substitutes. Number the string consecutively, starting at the end closest to a surrogate group. Remember that each substitute must have a number, and don't forget the prefixes: di, tri, tetra, etc. Solution Rule #1: Choose the longest substituted carbon chain that contains a functional group, doi:10.1351/goldbook.A00.222 Objectives After completing this sec It should be able to provide the correct IUPAC name for any given alkaline structure (Kekulé's, condensed or abbreviated). This gives the methyl groups the smallest possible number. The IUPAC names of hexane isomers are: A hexane B 2-methylpentane C 3-methylpentane D 2,2-dimethylbutane E 2,3-dimethylbutane Halogen substitutes are easily accommodated, using the names: fluorine (F-), chlorine (Cl-), bromine (Br-) and iodine (I-). A carbon attached to a functional group should have as few carbon numbers as possible. The suffix -ane identifies the molecule as an alkane, while the prefix identifies the carbon skeleton. To fulfill the second rule, we want to choose the numbering scheme that provides the lowest possible number of this substitute. They do not conduct electricity and do not polarize appreciably in electric fields. Consider alphabetical order; that is, after applying the first two rules mentioned above, make sure that your substitutes and/or functional groups are written in alphabetical order. A uniform variation of this type in a number of compounds is called homologous. The longest carbon chain has been highlighted in blue and consists of eight carbons. a) 1,3-dimethylbutane b) 4-ethylpentane c) 2-ethyl-3-methylpentane Answer a) The structure that can be drawn for the improper name is shown below on the left. When properly renumbered on the right), note that the most long chain is now 7 C  $\epsilon$   $\epsilon$   $\epsilon$   $\epsilon$   $\epsilon$   $\epsilon$   $\epsilon$   $\epsilon$  and since this Molecula is (In carbon 4), you can start the numbers of each end. If the halogen is linked to a simple alkyl group, an alternative name of "alkyl halide" can be used. For example, the correct name for the compound shown below is 3- methylheptane, not 2, ethylhexane. After taking into consideration the functional groups, any substitute must have the most possible carbon number. Rule # 3: After applying the first two rules, take into account the alphabetical order. The location of each chain is administered using a number. The names of multiple side chains (which do not count di-, tri-, etc. Alkanes of lower molecular weight tend to be gases and liquids, while the larger alquises are solid at room temperature. Each carbon Tomo participates in 4 chemical links. If we numered the chain on the left, bromine, the methyl group and chlorine would occupy the second, fifth and sixth positions, respectively. Given that the H / C relationship in these compounds is in a M ... Ximo, we llamas saturated (with hydrogen). (2006) When added to the water, they tend to decrease the entropy of the mixture or increase their level or order. One way to verify if the name you have given to you An alkane is reasonable is that it counts the number of carbon atoms implicit by the chosen name. When it renew it correctly (structure on the right), note that the long string is 6 c and, since this molá © Cula is symmetrical (between carbon 3 and 4), you can start the number S from each end. The prefixes) are given in alphabetical order before the name of the RAAZ chain. For example, isopentane, neopentane and n-pentane are names of branched shapes of the Alcane pentane. 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